

A SEARCH FOR PRESOLAR CHROMIUM-OXIDE (Cr_2O_3) GRAINS IN ORGUEIL. W.Fujiya¹, and N. Sugiura¹. ¹Department of Earth and Planetary Science, University of Tokyo. 7-3-1 Hongo, Tokyo 113-0033, Japan. E-mail: fujiya@eps.s.u-tokyo.ac.jp

Introduction: Isotopic anomalies of ^{54}Cr have been reported in bulk carbonaceous chondrites, especially least metamorphosed CI chondrites [1-4]. A Cr rich carrier(s) that has (have) ^{54}Cr excess was (were) found by step dissolution of carbonaceous chondrites [1,2,4]. Presolar grains are suggested to be potential carriers of the Cr isotopic anomalies. Actually, transmission electron microscopy (TEM) studies revealed the presence of Cr bearing-phases in Orgueil (CI): magnesiochromite (MgCr_2O_4), ureyite ($\text{NaCrSi}_2\text{O}_6$) and chromium-oxide (Cr_2O_3), in addition to chromite (FeCr_2O_4) [5]. Spinel grains (MgAlCrO_4 and FeCr_2O_4) were studied with Nano-SIMS and found to have oxygen and magnesium isotopic anomalies, and hence of presolar origin [6,7]. But the isotopic studies of chromium-oxide grains have not yet been performed. In this study, we report Cr isotopic composition of a chromium-oxide grain in Orgueil.

Experimental: A polished thin section of Orgueil was used in this study.

Cr rich grains were identified by EDS (energy dispersive X-ray spectroscopy) elemental mapping using a scanning electron microscope (SEM). The analyzed fields covered approximately $57600 \mu\text{m}^2$. We found about 20 Cr rich grains which range from submicron to a few micron in size. Subsequently we performed spot EDS analyses and detected one almost pure chromium-oxide grain with irregular and polygonal shape (Fig. 1) and some spinel grains.

This chromium-oxide grain was analyzed by the SIMS ion microprobe (ims-6f) in image acquisition mode using a resistive anode encoder (RAE). A primary beam of O^- with a diameter of $15 \mu\text{m}$ was used and the beam current was $\sim 0.1 \text{ nA}$. The O^- beam of 12.8 keV was irradiated on the sample surface of $\sim 70 \mu\text{m}$ in diameter and spatial resolution of images was $\sim 0.4 \mu\text{m}$. We obtained secondary ion images of ^{52}Cr , ^{53}Cr , ^{54}Cr , ^{56}Fe and ^{57}Fe . Their acquisition times were 600 seconds, respectively, except for ^{54}Cr whose acquisition time was 3000 seconds to reduce the counting error. The maximum count rate of the RAE was limited to $\sim 2 \times 10^4 \text{ cps}$. The ^{56}Fe count rate was the highest, mainly from the matrix of Orgueil. The field of view of the images was limited to $\sim 10 \mu\text{m}$ in diameter, in order to exclude extra signals of ^{56}Fe , resulting in high Cr count rate.

Isotopic fractionation of Cr and isobaric interference of ^{54}Fe were taken into account for estimating

^{54}Cr count rate. Fractionation of $^{54}\text{Cr}/^{52}\text{Cr}$ was estimated based on $^{53}\text{Cr}/^{52}\text{Cr}$ ratio. ^{54}Fe was estimated based on ^{56}Fe and ^{57}Fe . Dead time correction of the RAE detector was applied to ^{56}Fe .

It was assumed that Fe has the solar isotopic composition. We thought this assumption was valid because the grain was an almost pure chromium-oxide, so the Fe signals were attributed to the matrix around the grain.

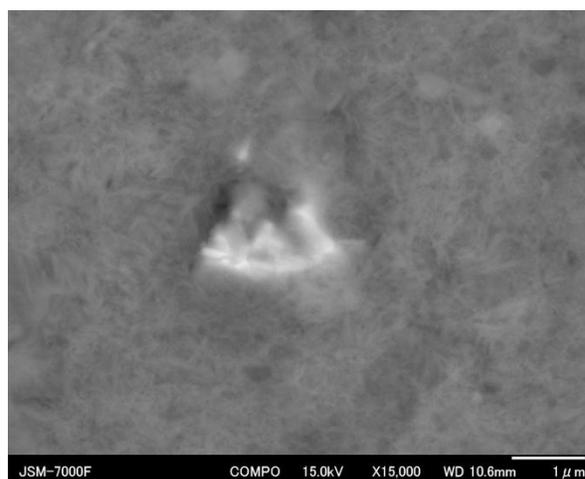


Figure 1. Backscattered FE-SEM image of a chromium-oxide grain (Cr_2O_3). This grain has irregular and polygonal shape.

Results and discussion: An example of obtained images is shown in Fig.2. In this case, the field of view is $\sim 20 \mu\text{m}$ in diameter. (Actually we measured isotopic composition with the field of view $\sim 10 \mu\text{m}$ in diameter.) The color shows counting rates in logarithmic scale. The Cr rich phase seems to be poor in Fe. The grain size appears to be a little large because of aberration of secondary ion optics.

For calculating isotopic compositions of Cr and Fe using the isotopic images by RAE, we integrated signals of the grain across 3×3 pixels. The maximum count rate of the RAE was 1.43×10^4 of ^{56}Fe . Considering dead time of the RAE detector was 4×10^{-6} second, dead time correction of 5.72% was applied to ^{56}Fe . Cr and Fe isotopic compositions of this grain were obtained to be $15 \pm 8\%$ for $\delta(^{53}\text{Cr}/^{52}\text{Cr})$ and $-5.6 \pm 2.3\%$ for $\delta(^{57}\text{Fe}/^{56}\text{Fe})$ (errors are 1σ). Mass dependent fractionations of Cr and Fe were calculated to be 30‰ for ^{54}Cr and 11.2‰ for ^{54}Fe . Total signals of the grain (3×3 pixels) were 1.6×10^5 counts for ^{52}Cr ,

8.5×10^4 counts for ^{56}Fe and 5.0×10^4 counts for ^{54}Cr . Therefore $\delta(^{54}\text{Cr}/^{52}\text{Cr})$ was calculated to be $-33 \pm 15\%$. This indicates the grain has small deficit or almost terrestrial value of $^{54}\text{Cr}/^{52}\text{Cr}$.

Oxygen and magnesium isotopic measurements of three spinel grains were previously performed by spot analyses with Nano-SIMS [6]. In the study, presolar spinel grains were identified but they had moderate depletions of ^{54}Cr or almost solar Cr isotopic compositions.

Grains in residues from treatments with acetic and nitric acids were studied by SIMS [8]. Some grains were found to show large isotopic anomalies of ^{54}Cr up to $\sim 800\%$. The chromium-oxide grain in our study and some spinel grains studied previously [6] are not potential carriers of very large anomalies of ^{54}Cr isotope. Ott et al. suggested that Cr rich grains with large grain size (high total ^{52}Cr count) had low ^{54}Cr anomalies [8]. In our study, the Cr rich grain is relatively large and the high total count of ^{52}Cr suggests that it belongs to the population of large grains of Ott et al. . Therefore, the lack of a large isotope anomaly is consistent with [8]. Although we have not detected plausible candidates of Cr rich phases with large excesses of ^{54}Cr so far, they are expected to be detected in the population of relatively small grains.

Stepwise dissolution of Orgueil showed that the largest excess ($\sim 100\epsilon$, 1ϵ is 1 part in 10^4) of ^{54}Cr did not occur in the most chemically resistant phases [1,2,4]. Spinel is one of the most chemically refractory phases. Therefore we think spinel grains are not the carriers with isotopic anomalies of ^{54}Cr , and chromium-oxide grains might be the candidates.

Stepwise dissolution also showed a deficiency of $\sim 6\epsilon$ in the most easily dissolved (acetic and nitric acids) fractions [1,2,4]. The fractions accounted for alteration phases and most ($>80\%$) of Cr in the whole rock. Isotopic composition of ^{54}Cr in bulk Orgueil appears to be a mixture of the materials that have excess or deficit of ^{54}Cr . Bulk Orgueil has ^{54}Cr excess of $\sim 1.5\epsilon$ and its Cr concentration is ~ 2500 ppm [3]. Considering the data above, if we assume that there are grains with very large ^{54}Cr excesses of 800% in Orgueil, the grain's concentration is calculated to be ~ 2.3 ppm.

It is noteworthy that ^{54}Cr is more homogeneously distributed in various phases in more metamorphosed chondrites [1,4]. Generally, presolar grains seem to have survived nebular processes and heating events in the early solar system. Therefore ^{54}Cr homogeneity in more metamorphosed chondrites appears to be somewhat strange. Anomalous Cr might be provided by late stellar input.

For future works, we will obtain more data about chromium-oxide and other Cr rich grains, and try to detect presolar Cr-bearing grains with ^{54}Cr excesses.

We expect Cr isotopic heterogeneity in meteorites to help our understanding of primordial materials of planetary bodies and constrain planetary formation processes in the early solar system.

References: [1] Rotaru M. et al. (1992) *Nature* 358, 465. [2] Podosek F. A. et al. (1997) *Meteorit. Planet. Sci.* 32, 617. [3] Shukolyukov A. & Lugmair G. W. (2006) *Earth Planet. Sci. Lett.* 250, 200. [4] Trinquier A. et al. (2007) *Astrophys. J.* 655, 1179. [5] Greshake A. et al. (1996) *Lunar Planet. Sci. XXVII*, 461-462. [6] Zinner E. et al. (2005) *Geochim. Cosmochim. Acta* 69, 4149. [7] Nittler C. M. et al. (2005) *68th Annual Meteorit. Society Meeting*, abstract #5208. [8] Ott U. et al. (1997) *Lunar Planet. Sci. XXVIII*, abstract #1278.

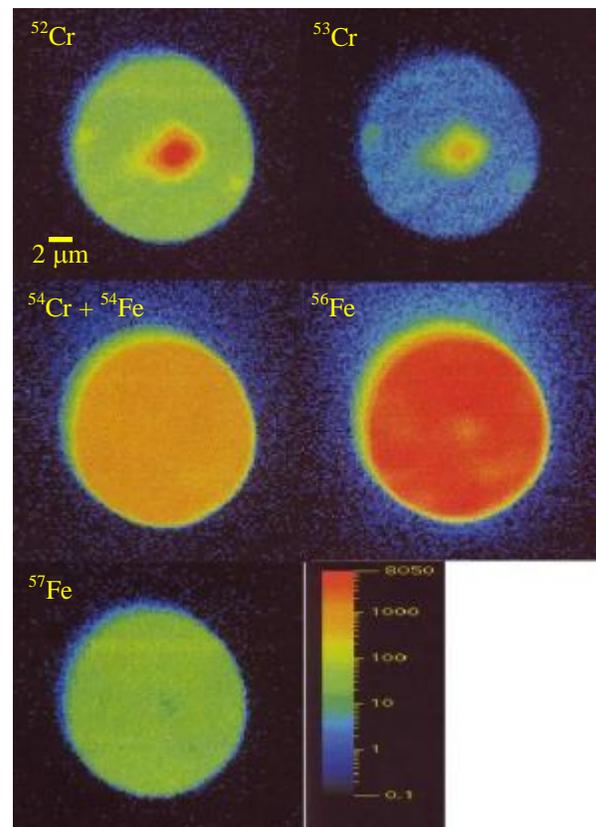


Figure 2. An example of obtained isotope images. The scale bar is 2 μm . The field of view is ~ 20 μm in diameter. Actually, we measured isotopic compositions with the field of view ~ 10 μm in diameter in order to exclude extra signals of ^{56}Fe in the matrix. For calculating isotopic compositions of Cr and Fe, we integrated signals of the grain across 3×3 pixels.